1	Prediction of the sweetening effect of Siraitia grosvenorii (luo				
2	han guo) fruits by two-dimensional quantitative NMR				
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16 ABSTRACT

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18 During the last decade, dried fruits of *Siratia grosvenorii* (luo han guo), also known as monk 19 fruit, have become popular food ingredients. Luo han guo extracts, which are promoted as non-20 caloric natural sweeteners, are now incorporated into dietary supplements, soft drinks, and 21 energy shakes. The compounds responsible for the sweetening effect are glycosylated 22 cucurbitane-type triterpenoids, the so-called mogrosides. However, of the more than 40 known 23 mogroside compounds, only $11-\alpha$ -hydroxy-mogrosides exhibit a sweetening effect, whereas the 24 other triterpenoids are non- or bitter-tasting. We have used two-dimensional quantitative NMR 25 to determine selectively the content of $11-\alpha$ -hydroxy-mogrosides in these dried fruits and thus 26 to predict their sweetening potential. Homonuclear (H,H COSY) and heteronuclear (HSQC) methods were developed, validated, and compared. Both techniques were found suitable for the 27 28 quality control of luo han guo fruits and extracts, the COSY experiment being advantageous 29 with regard to accuracy, precision, and limit of quantification.

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32 Keywords: *Siraitia grosvenorii*; Cucurbitaceae; luo han guo; mogrosides; HSQC; COSY;

33 qNMR; natural sweetener

34 **1. Introduction**

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36 Siraitia grosvenorii (Swingle) C.Jeffrey ex A.M.Lu & Zhi Y.Zhang, also referred to as 37 Momordica grosvenorii or Thladiantha grosvenorii, is a perennial vine of the Cucurbitaceae family (Pawar, Krynitsky, & Rader, 2016). Commonly known as "monk fruit" or "luo han guo", 38 39 S. grosvenorii is native to the southern part of China, where it has been used as a food ingredient 40 and herbal medicine for more than 300 years (Gong et al., 2019; Soejarto, Addo, & Kinghorn, 41 2019). S. grosvenorii is applied in typical ethnomedical treatments of ailments such as coughs, 42 lung congestion, and sore throats and was one of the first food homology species approved for 43 medical use in China (Gong et al., 2019). The plant was listed as a medicinal and edible species 44 by the China Ministry of Health in 1987 (Li et al., 2014). In the 1990s, the China Food and 45 Drug Administration approved luo han guo as a food sweetener and as a substitute for 46 sweeteners in health foods for patients with obesity and diabetes. Preparations of the dried fruits 47 have subsequently been marketed as sugar-free food additives and used in low-calorie health-48 promoting juices. Since 2011, extracts and preparations of S. grosvenorii have become 49 commercially available in the United States and are currently produced at large scale for utilization as table-top sweeteners (Fletcher, Pan, & Kinghorn, 2017; Pawar et al., 2016). 50 51 Having been awarded the GRAS ("generally regarded as safe") status in the U.S., S. grosvenorii 52 can now be considered the most popular natural sweetener after Stevia rebaudiana (Soejarto et al., 2019). 53

Like that of *S. rebaudiana*, the sweetening effect of *S. grosvenorii* constituents is reported to be up to 500 times as high as that of sucrose (Li et al., 2014; Pawar et al., 2016; Soejarto et al., 2019). However, in contrast to the sweet-tasting diterpenoids contained in stevia, the compounds responsible for the sweetening effect of luo han guo are glycosylated cucurbitanetype triterpenes. These compounds are characterized by a four-ring scaffold and various sugar 59 moieties at position C-3 and C-24 (Fig. 1). The major sugar present in mogrosides is D-glucose, 60 although L-rhamnose moieties have also been reported (Qing et al., 2017). The number of attached glucose units is expressed by Roman numerals in the compound name (rhamnoses are 61 62 indicated separately), with a reported maximum of seven sugar moieties (Qing et al., 2017). Depending on the nature of position C-11, which can be a carbonyl group, a methylene group, 63 64 or an α -hydroxy-methine group, in the scaffold, the compounds are referred to as 11-oxo-65 mogrosides (1), 11-deoxymogrosides (4), or mogrosides (2 and 3). The last-mentioned term is also used to circumscribe all of the luo han guo triterpenoids, however, only mogrosides with 66 the mogrol aglycone (2 and 3) possess the sweet taste. This means that the α -hydroxy-group is 67 68 crucial for the sweetness effect, as is the presence of at least three glucose moieties (Li et al., 69 2014).

70 So far, more than 40 cucurbitane-type triterpenoids have been isolated from S. grosvenorii, 71 of which 17 show the features necessary for exhibiting a sweet taste (Cicek, 2020). With regard 72 to the biosynthesis of mogrosides, the glucose moieties are successively attached to the 73 aglycone during the ripening process. Mogroside V (3), the major triterpenoid, appears 74 approximately 50 days after flowering and peaks after 70 days, whereas the mogroside IV class peaks as early as 40 days after flowering (Itkin et al., 2016; Pawar et al., 2016; Qing et al., 75 76 2017). Because of these variations over time and the overall complexity of the constituent 77 pattern, quantitative (chromatographic) methods for the quality control of luo han guo have 78 focused on one mogroside, namely mogroside V (3) (Deng, Liang, Yang, Liu, & Liu, 2013; 79 Makapugay, Nanayakkara, Soejarto, & Kinghorn, 1985) or a few constituents (Li et al., 2007; 80 Luo, Shi, Zhang, Qin, Guo, & Ma, 2016; Shen et al., 2014). In these studies, the amount of 81 mogroside V (3) has been found to vary between 5.8 and 12.9 mg/g dry fruit, whereas the 82 cumulated content of the major compounds has been reported to be between 8.8 and 18.5 mg/g (Luo et al., 2016; Makpugay et al., 1985; Shen et al., 2014). This contrasts with results obtained 83

from a colorimetric assay, which has determined the total content of mogrosides as being 38.2 mg/g dry fruit (Li et al., 2014). However, neither colorimetric nor chromatographic methods have so far accomplished a determination of the total amount of $11-\alpha$ -hydroxy-mogrosides or an analysis of the sweetening potential of luo han guo at a molecular level.

In the present work, we investigated the fruits of *S. grosvenorii* with two-dimensional NMR spectroscopy by using its characteristic measuring principle, the detection of specific structural features. We hypothesized that signals exclusively resulting from $11-\alpha$ -hydroxy-mogrosides could be used for the quantification of this particular compound class. This strategy should allow the selective determination of the compounds of interest and consequently the prediction of the sweetening effect of the ripe fruits.

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95 2. Materials and methods

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99 Dried fruits of Siraitia grosvenorii (Golden Diamond, Lot B02/2549) were obtained from 100 the Go Asia supermarket (Hamburg, Germany). 1,2,4,5-Tetrachloro-3-nitrobenzene TraceCert for quantitative NMR (Lot BCBR7216V, 99.86%), LC-MS grade formic acid, Diaion HP-20, 101 102 and Sephadex LH-20 were purchased from Sigma Aldrich Co., St. Louis, MO, USA. Silica gel 103 (40-63 μ m) for column chromatography, TLC plates (silica gel 60 F₂₅₄), acetonitrile and water 104 (both of LC-MS grade), gradient grade methanol, and other (analytical grade) solvents were 105 obtained from VWR International GmbH, Darmstadt, Germany. Water used for isolation was 106 doubly distilled in-house. Solid phase extraction (SPE) columns (Chromabond SB 3 mL/500 107 mg, Chromabond C18 6 ml/1000 mg) were obtained from Macherey-Nagel GmbH and Co. KG, Düren, Germany. Dimethyl sulfoxide- d_6 (99.80%, Lot S1051, Batch 0119E) and pyridine- d_5 108

^{97 2.1.} Plant material and chemical reagents

109 (99.50%, Lot Q1301, Batch 1116) for NMR spectroscopy were purchased from Euriso-top
110 GmbH, Saarbrücken, Germany, and conventional 5 mm NMR sample tubes were obtained from
111 Rototec-Spintec GmbH, Griesheim, Germany.

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113 2.2. General experimental procedures

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Thin layer chromatography was performed using *n*-butanol–ethanol–water (40-10-15) as 115 116 the eluent and vanillin sulfuric acid as the spraying reagent. Flash chromatography was carried 117 out with a Buchi PrepChrom C-700 chromatograph and FlashPure Ecoflex (50 µm irregular, 118 10 g) silica gel cartridges (Büchi Labortechnik GmbH, Essen, Germany). Preparative MPLC 119 was performed with the same instrument but by using a Buchi PrepChrom C18 column (250 \times 120 30.0 mm, 15 m particle size). Semi-preparative HPLC was accomplished using a Waters 121 Alliance e2695 Separations Module with an Alliance 2998 photodiode array, and a WFC III 122 fraction collector (Waters, Milford, MA, USA) using a VP Nucleodur C18 column (250 × 10 123 mm, 5 µm particle size, Macherey-Nagel GmbH and Co. KG, Düren, Germany). UHPLC 124 analyses were performed on a VWR-Hitachi Chromaster Ultra RS equipped with a 6170 binary pump, 6270 autosampler, 6310 column oven, 6430 DAD, and VWR 100 evaporative light 125 126 scattering detector (VWR International GmbH, Darmstadt, Germany). LC-MS analyses were 127 carried out on a Shimadzu Nexera 2 liquid chromatograph connected to an LC-MS triple 128 quadrupole mass spectrometer using electrospray ionization (Shimadzu, Kyoto, Japan). A 129 Phenomenex Luna Omega C18 column (100×2.1 mm, 1.6μ m particle size, Phenomenex, 130 Aschaffenburg, Germany) was employed for the analysis of extracts, fractions, and pure 131 compounds during isolation and method development. Quantification of mogroside V (3) was 132 performed on a Nucleodur C18 Pyramid (250 × 4.6 mm, 5 µM, Macherey-Nagel, Düren, Germany) by triplicate injections of both, the mogroside reference and the luo han guo sample. 133

135 2.3. NMR spectroscopy

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137 NMR spectra were recorded using a Bruker Avance III 400 NMR spectrometer (Bruker Biospin, Rheinstetten, Germany) operating at 400.33 MHz for the proton channel and at 100.66 138 MHz for the 13 C channel by means of a 5 mm PABBO broad-band probe with a z gradient unit. 139 140 Measurements were performed at 293 K; the temperature in the probe head was calibrated with 141 a methanol- d_4 solution and regulated within 0.1 K during the measurements. For each sample, 142 automatic tuning and matching of the probe was performed, as was automatic shimming of the on-axis shims (z to z^5). The automatic receiver gain adjustment mode afforded the maximum 143 144 gain value in all measurements. Bruker Topspin software 3.6.0 was employed for spectra 145 recording.

146 Characterization of the purified sample in pyridine- d_5 and DMSO- d_6 was accomplished by 147 use of conventional full spectral range routine parameters and pulse programs for structure 148 elucidation.

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150 2.3.1 Quantitative H,H COSY measurements

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152 The non-phase-sensitive pulse program cosygpmfqf (with gradient pulses) and double 153 quantum filter selection via the gradient ratio supplied by the manufacturer were used for 154 correlated spectroscopy (COSY) measurements. Spectral width was set to 3.10 ± 2.40 ppm (a 155 range of 1921 Hz) in both dimensions, with 1024 data points being collected in F2 and 128 156 increments in the indirect (F1) dimension. Acquisition time was calculated to be 0.267 s, and 157 the interscan-delay was 3 s. With two scans collected per increment and 16 dummy scans, the 158 total measuring time amounted to 14 min and 50 s ('2-scan-COSY'); with eight scans, about 57 159 minutes were required to record the spectrum ('8-scan-COSY').

160 The raw data were processed as follows: zero filling to an $8k \times 8k$ matrix, then 161 multiplication with a sine function before two-dimensional Fourier transformation and 162 magnitude calculation in F2, with automatic baseline correction in both dimensions and 163 symmetrization of the resulting spectrum. Integration borders were set uniformly by using the 164 same integration file with a range of 3.762 to 3.604 ppm for F2 and 1.792 to 1.612 ppm in F1.

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166 2.3.2. Quantitative band-selective ¹H, ¹³C HSQC measurements

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Band-selective heteronuclear single quantum correlation spectroscopy (bs-HSQC) measurements for quantification were performed using the phase-sensitive *shsqsctgpsi2.2* pulse program of the Bruker pulse program library with a band-selective shaped ¹³C refocusing pulse and ¹³C GARP decoupling. The ¹H spectral range was (4.50 ± 5.05) ppm, spectral range 4006 Hz, whereas the ¹³C spectral range was (77.2 ± 5.0) ppm, corresponding to 1006 Hz.

173 The shape form Q3.1000 was chosen for the selective refocusing pulse and, in order to achieve selective excitation over the frequency range of 1000 Hz (10 ppm), the length of the 174 pulse was determined to 3448 µs with a power of 0.0986 W for the probe head in use. For the 175 ¹H channel, 1024 data points were collected, whereas for the ¹³C channel, 64 data points were 176 177 set. The acquisition time was 0.128 s, and the inter-scan delay was set to 3.15 s. The relevant delays in the pulse program were set corresponding to a coupling constant of ${}^{1}J_{CH} = 135$ Hz. 178 With 16 dummy scans and 16 scans per increment, the total acquisition time for this HSQC 179 180 experiment was about 57 minutes.

181 The raw data were processed as follows: zero filling to a $4k \times 4k$ matrix and then 182 multiplication of both dimensions with a squared sine function before two-dimensional Fourier 183 transformation and phase correction. Phase correction parameters of the previous experiment 184 were used and were carefully visually controlled and manually adopted for the relevant cross 185 peak in both dimensions for each measurement. After automatic baseline correction in both dimensions, integration borders were set uniformly by using the same integration file with a range of 3.77 to 3.60 ppm in F2 (¹H) and 78.10 to 76.26 ppm in F1 (¹³C, bs-HSQC_{wide}). Additionally, an alternate integration range (bs-HSQC_{small}) was used with much narrower ranges of 3.77 to 3.60 ppm in F2 (¹H) and 77.37 to 76.76 ppm in F1 (¹³C).

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191 2.4. Extraction and isolation

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193 A total of 600 g powdered dried fruits was exhaustively extracted with methanol by maceration, and the solvent was evaporated under reduced pressure to yield 114.3 g crude 194 195 extract. The crude extract was dissolved in 300 mL water and chromatographed over Diaion 196 HP-20, with subsequently elution with water, methanol 30%, methanol 80%, and methanol 197 100%. The methanol 80% fraction (17.5 g) was subjected to silica gel column chromatography 198 by using dichloromethane-methanol-water in three portions of 815 mL with the following 199 ratios: 8:2:0.2, 7:3:0.5, and 6:4:1. Of the resulting eight fractions (A–H), fraction G (8.70 g) 200 was chromatographed with Sephadex LH-20 to give another five fractions (G1–G5). Fraction 201 G3 (5.70 g) was subjected to preparative MPLC by applying a solvent mixture of 0.1% formic 202 acid in water (solvent A) and methanol (solvent B) with the following gradient: 38% B to 45% 203 B in 60 min, and to 95% B in 30 min. Of the resulting five fractions (G3A–G3E), fraction G3B 204 (716 mg) was chromatographed using silica gel flash chromatography and a gradient of 205 dichloromethane-methanol-water (8:2:0.2 to 7:3:0.5) to give six fractions. The third fraction 206 (211 mg) was separated by semi-preparative HPLC by using 0.025% formic acid in water 207 (solvent A) and acetonitrile (solvent B) with a gradient of 25% B to 30% B in 30 minutes, a 208 flow rate of 2 mL/min, and an oven temperature of 35°C, to give 26.9 mg compound 1 and 27.5 209 mg compound **3**. Fraction G3C (371 mg) was purified using normal phase solid phase extraction 210 to give another 113.8 mg compound 3. Fraction G3D (242 mg) was subjected to semi-

211	preparative HPLC by using 0.025% formic acid in water (solvent A) and acetonitrile (solvent
212	B) with a gradient of 20% B to 25% B in 30 minutes, a flow rate of 3 mL/min, and an oven
213	temperature of 35°C, to give 27.5 mg compound 2 and 32.3 mg compound 4 .

215 *2.5. Quantitative analysis*

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A total of 1.00 g dried and ground fruits was extracted three times with 50 mL methanol-217 218 water 80% (v/v) by using ultra-sonication for 15 min each. After filtration, the solution was 219 evaporated to dryness. The residue was re-dissolved in 3 mL methanol-water 20% (v/v) and 220 subjected to an equilibrated SPE C-18 column (1 g, 6 mL). After the column had been washed 221 with three column volumes of methanol-water 20% (v/v), the desired compounds were eluted 222 with three column volumes of methanol (100%). The combined methanol fractions were 223 evaporated to dryness, re-dissolved in 600 μ L DMSO- d_6 , and transferred to NMR sample tubes. 224 For calibration, 26.4 mg mogroside V (3) were dissolved in 1000 μ L DMSO- d_6 and diluted to 225 the desired concentrations with the same solvent. Purity of the compound was assessed using 226 1,2,4,5-tetrachloro-3-nitrobenzene (6.20 mg) as the internal standard.

EVALUATE: For the quantification of $11-\alpha$ -hydroxy-mogrosides, calibration solutions 1-5 with concentrations of 20.34, 15.26, 8.48, 3.39, and 1.70 mmol/L were prepared and analyzed with bs-HSQC and H,H COSY measurements. For the latter experiment, two different parameter sets were investigated by using either 2 or 8 scans and the resulting acquisition times of 15 min or 57 min, respectively.

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233 2.6. Method validation

235 The developed method was validated for linearity, repeatability, precision, accuracy, and 236 limit of quantification. Evaluation of linearity was achieved by establishing calibration curves 237 over a range of at least 80 to 120% of the measured concentrations. Therefore, 5-point 238 calibration curves were created and expressed as linear functions. In order to define a limit of 239 quantitation, a solution of 1.00 mmol/L mogroside V (3) was prepared. Precision and repeatability were assessed using samples of luo han guo. Repeatability measurements were 240 241 conducted six times on the same sample. For the determination of precision, six different 242 samples were prepared and measured once within one day (intra-day precision) or on two 243 different days (inter-day precision). Determination of accuracy was accomplished by spiking 244 the samples with various amounts of standard solution as follows: 245 high spike (120%): $150 \,\mu\text{L}$ sample + $450 \,\mu\text{L}$ calibration solution 2 246 247 medium spike (100%): 150 μ L sample + 225 μ L calibration solution 2 248 + 225 µL calibration solution 3 249 low spike (80%): $150 \,\mu\text{L}$ sample + 450 μL calibration solution 3 250 All obtained values were expressed as % recovery by using the mean value of the inter-day 251 252 measurements (N=12) for the sample concentration. Additionally, an overall mean value was 253 calculated from the inter-day measurements of all four experimental results (N=48), and the 254 deviations of the results obtained from the four methods were compared. 255 256 3. Results and discussion 257 258 3.1. Isolation of standard compounds and revision of NMR data 259

260 As a first step, standard compounds were isolated for method development and eventual 261 validation. At least one compound with an α -hydroxy-group in position C-11 and one compound without this structural feature were required. Therefore, we focused on the isolation 262 263 of mogroside V (3) and 11-oxo-mogroside V (1), the two most abundant mogrosides in the ripe 264 fruits of S. grosvenorii (Shen et al., 2014). The isolation steps followed in principle the 265 procedure described by Li et al. (2006) for the isolation of mogroside V (3). The isolation 266 process was monitored by LC-DAD-ELSD and LC-MS using the method described by Qing et 267 al. (2017).

268 All in all, four cucurbitane glycosides were obtained and identified as 11-oxo-mogroside V (1), mogroside IVe (2), mogroside V (3), and 11-deoxymogroside V (4) by using MS and NMR 269 270 experiments and comparisons with literature data (Akihisa et al., 2007; Chaturvedula, & 271 Prakash, 2011; Kasai et al., 1989; Niu et al., 2017; Prakash, & Chaturvedula, 2014; Qing et al., 272 2017). However, the reported NMR shift values for mogroside V (3) and 11-oxo-mogroside V 273 (1) (Chaturvedula, & Prakash, 2011) differed significantly from our own NMR measurements 274 and from the NMR data of similar compounds (Akihisa et al., 2007; Niu et al., 2017). Revised 275 NMR data for the two major components mogroside V (3) and 11-oxo-mogroside V (1) are 276 given in Tables S1 and S2 together with those of the other two isolated compounds. Moreover, 277 we reassigned shift values for five signals of mogroside IVe (2), which had not been 278 unambiguously assigned by the time of its structure elucidation (Kasai et al., 1989). 279 Additionally, we added missing values for the sugar moieties of the compound also referred to 280 as mogroside IV in older references. The NMR spectra of all four compounds are depicted in 281 Figs. S1-S30.

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283 *3.2. Method development*

287 In our quest to identify and quantify all $11-\alpha$ -hydroxy-mogrosides, we first analyzed 288 significant signals of the terpenoid moiety by comparing mogroside V (3) and 11-oxo-289 mogroside V (1). Several prerequisites had to be fulfilled for the selective quantification of the 290 $11-\alpha$ -hydroxylated species, in addition to structures with either a methylene group or a carbonyl 291 function in the respective position (see Fig. 1). Because of the severe overlap of many aliphatic 292 signals of the terpenoid ring system, the high frequency shifted methine proton next to the 293 hydroxyl group (H-11) displays a unique feature of sweet-tasting mogrosides. In the one-294 dimensional ¹H NMR spectrum recorded in pyridine- d_5 , this signal unfortunately overlapped 295 several methine signals of the glucose moiety. Thus, direct integration was not possible. HSQC 296 experiments showed similar results, because the shift value of C-11 was also within the bulk of 297 glucose signals. However, selectivity was achieved when the nuclei in the neighbourhood of 298 this proton were considered. H-11 showed coupling with the two protons in position 12, both 299 having a chemical shift of 2.19 ppm (Table S2). This coupling appeared as a clearly separated 300 cross peak in the two-dimensional H,H COSY spectrum for the 11-a-hydroxy-mogrosides and 301 showed no overlap with any sugar signals (Fig. S15, 4.18 ppm/2.19 ppm).

302 NMR spectra for compounds 1 and 3 were additionally recorded in DMSO- d_6 , because of its 303 potent solvency for both terpene glycosides and the crude extracts. Moreover, dimethyl 304 sulfoxide is a far more economic NMR solvent. Again, 1D proton NMR spectra did not provide 305 sufficient resolution for the H-11 methine signal (Fig. S19), which now overlapped with three 306 other diasterotopic protons of the terminal glucose methylene groups GII, GIV, and GV (Fig. 307 1). However, this time not only H,H COSY measurements showed good resolution (Fig. S21), 308 but also HSQC experiments clearly separated the CH-11 cross peak from all other glucose 309 signals (Fig. S22). Consequently, DMSO- d_6 was chosen as the NMR solvent, and the cross peak of the CH-11 correlation at 3.67 ppm/76.7 ppm (HSQC) and the cross coupling of the H11 methine signal with the H-12 methylene protons at 3.67 ppm/1.67 ppm (H,H COSY) were
selected for further method development.

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314 3.2.2. Extraction and sample preparation

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316 For the extraction of the dried fruits, pure methanol (Shen et al., 2014) or methanol-water 317 (80% v/v) (Qing et al., 2017) have been described as suitable solvents for analytical purposes, 318 with a ratio of 1 g powdered plant material to 100 mL of solvent being applied. In our study, 319 the use of methanol resulted in a yield of approximately 24% (w/w) of crude extract calculated 320 based on the amount of dried fruits, whereas the methanol-water mixture delivered 36% (w/w) 321 of crude extract. The higher mass fraction of the methanol-water extract was expected, and a 322 higher concentration of α-hydroxy-mogrosides was observed. LC-MS analysis of the crude 323 extracts revealed that both solvents efficiently extracted mogrosides with two, three, four, or 324 five sugar moieties, but only the methanol-water mixture yielded hexaglycosidic mogrosides. 325 Thus, methanol-water 80% (v/v) was used for subsequent extractions.

In order to increase the concentration of α -hydroxy-mogrosides and to reduce NMR 326 327 measurement time, the crude extract was further purified by C-18 solid phase extraction. 328 Therefore, various methanol-water mixtures (0, 20, 50, and 80%) were evaluated for their 329 ability to wash polar analytes off the column without eluting the compounds of interest. LC-330 MS analysis confirmed that, with methanol (20%), a large number of unwanted constituents 331 were removed before elution of the α -hydroxy-mogrosides with pure methanol. This procedure 332 significantly increased the concentration of the α -hydroxy-mogrosides, leading to intense 333 signals in both the H,H COSY and HSQC experiments.

337 Utilizing the standard pulse programs delivered by the spectrometer manufacturer, we 338 developed a method for the measurement and integration of several H,H COSY spectra. H,H 339 COSY measurements provide the option of the selective detection of one special coupling via 340 the corresponding cross peak with proton excitation and proton detection at 3.67 ppm/1.67 ppm. 341 Among several H,H COSY variants, double quantum filtered experiments involving the 342 application of gradient pulses and thereby the suppression of singlets and artefacts are 343 considered extremely robust (Shaw, Salaun, Dauphin, & Ancian, 1996). We employed an 344 established simple non-phase-sensitive measurement allowing easy integration of parts of the 345 2D plot. Because only one signal region was of interest, we used a small spectral width, 346 observing the aliphatic region in the range of 0.7 to 5.5 ppm and thus reducing measurement 347 time (Fig. 2). The crucial experimental parameter to be investigated for quantitative analysis is the scan repetition time, as this parameter is responsible for the length of the NMR experiment. 348 349 It has to be sufficiently long to allow complete relaxation of the protons. We investigated this parameter by running H,H COSY experiments with various delays between 0.5 and 20 s for the 350 351 standard compound (3) and the extract after sample preparation. No significant differences of 352 the integrals of the cross peak of H-11/H-12 in the two cases were observed in the scan 353 repetition time range of 2 s to 20 s (data not shown). Thus, a setting of 3 s was used for all H,H 354 COSY experiments in this study. Two measurements were made: a fast 2-scan-COSY (15 min) 355 and an 8-scan-COSY (57 min) with improved signal to noise ratio. During the evaluation of the 356 processing and integration parameters, we obtained most reproducible results by improving the 357 resolution by zero filling and, after automatic baseline correction on both dimensions, by 358 symmetrizing the H,H COSY spectrum (see section 2.3.1.).

362 The other nucleus in the neighbourhood of the H-11 methine proton is the carbon atom at 363 position 11. This one-bond coupling is responsible for the cross signal in the HSQC spectrum 364 at 3.67 ppm/76.7 ppm (Fig. 3). Here, protons are excited, the polarization is transferred to the 365 carbon atom and back to the proton again, which are then detected in a sensitive manner 366 (Schleucher et al., 1994). We used this approach as a second quantitative method and observed 367 that no actual overlap occurs with any glucose signals. For the optimization of this HSQC 368 experiment, we investigated once more the effect of the relaxation time and found no significant 369 difference between delays of 2 to 20 s (data not shown). Therefore, a value of about 3 s was 370 also chosen for this experiment. We drastically reduced the number of increments for the carbon 371 frequency by working with band-selective excitation of only a range of 10 ppm (Girreser, 372 Ugolini, & Cicek, 2019). Furthermore, delays in the pulse program were adapted to the coupling 373 constant of this methine unit (Cicek, Ugolini, & Girreser, 2019), which we determined at 135 374 Hz. Finally, we compared the traditional recording of each increment with non-uniform 375 sampling (Jaravine, Ibraghimov, & Orekhov, 2006) and obtained more reproducible results 376 with traditional sampling (data not shown). In this manner, we created a method requiring a 377 measuring time of about 57 min. Again, we paid attention to the data processing by zero filling 378 of the 2D matrix, automatic baseline correction, and the selection of a proper integration range. 379 In this case, we selected a small and a wide integration range and compared the results of the 380 different ranges (see section 2.3.2).

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382 *3.3. Method validation*

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384 3.3.1. Linearity and quantitation limit

For the validation of linearity, calibration curves in four different modes were established over a concentration range of 1.70 to 20.34 mmol/L, with mogroside V (**3**) as an external calibration standard (Table 1). Purity of the compound was checked by using a certified reference standard (1,2,4,5-tetrachloro-3-nitrobenzene) and integration of the H-6 in the cucurbitane scaffold, plus the H-6b of the G_I glucose moiety, by 1D proton NMR. The purity of mogroside V (**3**) was determined to be 99.09%.

All four calibration curves showed acceptable coefficients of determination, with values between 0.9965 and 0.9986 (Table 1). In the HSQC experiments, both modes of integration yielded almost the same slopes and only differed in their offsets, which were naturally higher for the wide integration range. The differences between the two COSY experiments were characterized by a significantly higher slope for the 8-scan-COSY, which furthermore showed a negligible offset.

398 Because of the use of integration fields instead of the threshold integration, a limit of 399 quantitation had to be defined. Therefore, a solution of 1.00 mmol/L was prepared and 400 determined under the four measurement/integration modes. Whereas, with the 8-scan-COSY, 401 this concentration was determined within 50% of the actual value, the other three modes did 402 not show sufficient accuracy at this level. For these three modes, the limit of quantitation was 403 defined as 1.70 mmol/L (Table 1). When these values were calculated as mogroside V (3), as 404 we did in the quantitative assays (Table 2), they corresponded to 1.3 mg/mL (1.00 mM) or 2.2 405 mg/mL (1.70 mM), respectively. Because 1.00 g luo han guo was applied in the assays, the 406 same concentrations are present per gram of dried fruit.

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408 3.3.2. Precision and repeatability

410 HSQC measurements showed similar repeatability, with a relative standard deviation of 411 3.38% for the wide integration mode and 3.25% for the small integration parameters (Table 2). 412 Moreover, in terms of precision, both variants were comparable, with RSD values between 413 5.45% and 6.35%. H,H COSY measurements showed better repeatability than the HSQC 414 experiment, with RSD values of 1.95% for the 2-scan-COSY and 1.89% for the 8-scan-415 experiment. The latter experiment additionally displayed excellent values during the precision 416 measurements, with relative standard deviations of 1.66 to 1.73%. The 2-scan COSY, in 417 contrast, showed RSD values similar to those of the HSQC experiment, with relative standard 418 deviations of 4.43 to 5.37%. Interestingly, no significant differences were observed between 419 the intra-day and inter-day measurements, indicating the robustness of the developed NMR 420 methods (Table 2).

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422 3.3.3. Accuracy and comparability

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424 Spiking experiments were conducted for the determination of accuracy (Table 3). Various 425 calibration solutions were used to achieve concentrations of 80, 100, and 120% of the assay concentration (mean values of the precision measurements). With regard to the HSQC 426 427 experiments, both integration methods revealed good recovery rates and showed slightly better 428 recoveries for the small integration range. H,H COSY measurements also resulted in excellent 429 recovery rates, with maximum deviations of 4.9% for the 2-scan-experiment and 2.8% for the 430 8-scan-COSY. In addition, the mean values obtained from the precision measurements were 431 compared with each other and showed usual deviations of not more than 3% (Table 3). 432

433 3.3.4. Specificity and selectivity

435 The total content of $11-\alpha$ -hydroxy-mogrosides can be determined from both the HSQC 436 experiment and the H,H COSY measurements. Moreover, these value-determining compounds 437 can be quantified among additional non-sweet-tasting derivatives, e.g., 11-oxo-mogrosides or 438 11-deoxy-mogrosides. Therefore, the present methods can be rendered specific and selective at 439 the same time. The selective determination of the sweetening principle can be achieved in two 440 ways: either by direct observation of the CH-11 methine unit (HSOC) or, alternatively, through 441 indirect (H,H COSY) measurement via the coupling to the methylene group in the 442 neighbourhood (CH₂-12). This highlights the advantage of NMR spectroscopy over other 443 analytical methods, namely its characteristic measuring principle and the possibility of quantifying the structural features necessary for the biological effect. 444

445

446 *3.4. Quality control*

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448 In the above, we have presented two experiments for the quality control of S. grosvenorii. 449 The amount obtained for the $11-\alpha$ -hydroxy-mogrosides was determined as being 15.2 to 16.2 450 mg/g dried fruit (calculated as mogroside V). In comparison, the amount of mogroside V (3) 451 was determined as 9.2 mg/g dried fruit by using the method of the Food Chemicals Codex (Figs. 452 S31 and S32) (FCC 11, 2018). This value is in accordance with the average amounts described 453 in the literature for mogroside V (3), namely 9.5 mg/g (Li et al., 2014) or 10.5 mg/g (Pawar et 454 al., 2013), respectively. However, the significantly lower value compared with our assays 455 indicates the large amounts of additional constituents that contribute to the sweetening effect 456 of luo han guo and that are not quantified with the chromatographic methods reported to date.

457 We investigated two variants in the two experiments. For the HSQC measurements, we used 458 different integration ranges (bs-HSQC_{wide} and bs-HSQC_{small}), whereas for the H,H COSY 459 experiments different scan numbers were applied (2-scan-COSY and 8-scan-COSY). For the

HSQC experiment, a selective pulse program decreased the ¹³C spectral range to 10 ppm and 460 461 thus significantly reduced the measurement time. Validation of the four resulting methods 462 according to the ICH guideline Q 2 R(1) have revealed similar values for the two HSQC variants 463 thus indicating the applicability of the two integration methods. For the H,H COSY 464 measurements, a 2-scan experiment was investigated for the fast quantification (15 min) of 11-465 α -hydroxy-mogrosides, in addition to an 8-scan variant, resulting in the same measurement time as in the HSQC experiments (57 min). The 2-scan-COSY revealed similar validation results as 466 467 the HSQC experiments, whereas the 8-scan-COSY was superior in terms of accuracy, precision, 468 and limit of quantitation.

Furthermore, all methods were able selectively to quantify the total content of the value-469 470 determining $11-\alpha$ -hydroxy-mogrosides. Because the assay is based on molarity (and the content is subsequently referenced to mogroside V), the method can also be applied to unripe fruits, 471 472 and the sweetening potential can be predicted and estimated at an early stage of cultivation. 473 Another issue when dealing with luo han guo is the possible dilution of extracts with extracts 474 of unripe fruits (Pawar et al., 2013). Although not being intended for the detection of 475 adulterations, our method can also be performed to address this problem by using the cross 476 correlations of the H-6 protons of the non-terminal glucose moieties G_I and G_{III} (Fig. 1). These 477 cross peaks can be found in the H,H COSY spectrum at 3.57/3.90 ppm for G_I and at 3.40/4.11478 ppm for G_{III} and have a ratio of 2:1 with regard to the H-11/H-12 cross peak of pure mogroside 479 V (3) (Fig. 2). In luo han guo samples, this ratio is approximately 2.75:1. The higher ratio results 480 from the remaining mogrosides, which obviously show similar grades of glycosylation. 481 However, in the case of dilution with unripe fruits, the ratio would be far below 2:1, because of 482 the increased presence of mogrosides with only two terminal sugar moieties (one at C-3 and 483 one at C-24) (Itkin et al., 2016). Thus, the characteristic signals of the non-terminal glucose 484 moieties would be significantly reduced (or absent in the case of complete falsification). The

485 detection of adulteration would also be feasible with the HSQC experiment by using the 486 respective C-H correlations. Here, a ratio of 0.6:1 was observed in our experiments for 487 mogroside V (**3**) and 0.85:1 for luo han guo samples. However, in the band-selective mode, the 488 13 C spectral width would have to be increased resulting in longer measurement times.

489 Analysis time is an important issue when dealing with quality control methods and is one 490 of the reasons that two-dimensional NMR methods are readily overlooked. However, in the 491 present study and with regard to the application of band-selective approaches for the HSQC 492 experiment, we have demonstrated that even heteronuclear techniques deliver precise and 493 accurate results within reasonable measurement times. For example, in the HPLC assay for the 494 mogroside V (3) content of the Food Chemicals Codex, the last peak in the extract eluted after 495 more than 50 minutes (Fig. S32). Thus, a measurement time of 60 minutes was chosen, which 496 corresponds to the measurement times of three of our methods. Chromatographic separation of 497 extracts will always need several minutes of elution time, especially for more complex extracts 498 such as the extracts of luo han guo. Elution times can (to some extent) be decreased using 499 smaller particle sizes, selective extraction, or mass spectrometric detection methods. Deng et 500 al. (2013) have used micelle-mediated cloud-point extraction and quantified mogroside V (3) 501 in 20 minutes. However, the relatively low accuracy and precision values and the extraction 502 yield of only 80% in their investigation are major shortcomings for the quality control of luo 503 han guo. In two other studies, LC-MS was applied in the multiple reaction monitoring (MRM) 504 mode and quantified five and eight mogrosides, respectively (Shen et al., 2014; Luo et al., 505 2016). The analysis time of the two methods was 12 and 10 minutes and thus in the range of 506 the 2-scan-COSY experiment. Moreover, in terms of accuracy and precision, the two LC-MS 507 methods are comparable with the 2-scan-COSY, showing slightly better precision values and 508 somewhat lower recovery rates. The linear range of MRM methods usually is in the nanomolar 509 scale and obviously results in lower limits of quantification. These low LoQs render MRM

510 methods an ideal tool for trace analysis. For quality control assays, however, the LoQ is not a 511 crucial parameter and is not even requested in the respective ICH guideline, because the assay 512 concentration is determined during method development. Nevertheless, broader quantitation 513 ranges paired with selectivity allow easier method adaption to other purposes, e.g., analysis of 514 dietary supplements. Both characteristics are given for the NMR methods described herein. In 515 contrast to all previously reported assays, all (rather than just a few) of the sweet-tasting 11-α-516 hydroxy-mogrosides have been quantified in the ripe fruits of *Siraitia grosvenorii*.

517

518 **4. Conclusion**

519

In this study, two-dimensional quantitative NMR was successfully applied for the analysis 520 521 of S. grosvenorii fruits. The determination of the total content of sweet-tasting mogrosides was 522 accomplished by using NMR signals resulting from the $11-\alpha$ -hydroxy-group that occurs in this 523 particular compound class and that is the crucial structural feature for the sweetening effect. In 524 contrast to our previous work in which we quantified the total amount of a compound class by 525 one specific NMR signal (Çiçek, Girreser, & Zidorn, 2018), only the compounds relevant for 526 the biological effect were determined this time. The selective quantification of bioactive 527 compound (sub)classes is not only a valuable tool for the quality control of food ingredients or 528 herbal medicines but can also be applied to the screening of extracts during, for example, drug 529 discovery. Furthermore, the present study provides another example for the advantage of 530 quantitative NMR spectroscopy in natural products analysis in which several constituents 531 contribute to a biological effect.

532

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- 537
- 538





542 Fig. 1: Chemical structures of 11-oxomogroside V (1), mogroside IVe (2), mogroside V (3),

543 and 11-deoxymogroside V (4). Exact stereochemistry can be retrieved from compounds 2 and

- 544 **3**.
- 545

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546

547 Fig. 2: H,H COSY diagram of luo han guo sample recorded in DMSO- d_6 in the region of 0.7 to

548 5.5 ppm. The signal used for quantitative NMR is highlighted in red.



550

Fig. 3: Complete spectral range HSQC diagram of luo han guo sample recorded in DMSO- d_6 and band-selective HSQC diagram in the region of 3.3 to 4.0 ppm (¹H) and 67 to 87 ppm (¹³C) (top left, only signals in the ¹³C-range from 72.2 to 82.2 ppm are detected). Signal used for quantitative NMR is highlighted in red.

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- 556 557

558 Table 1

Regression equations, coefficients of determination (R^2) , and limits of quantitation (LoQ) 559 obtained from various NMR measurements (y: 2D NMR integral in arbitrary units, x: 560 561 concentration in mmol/L)

	Measurement		Regression equation	R^2	LoQ			
	bs-HSQC (wide in	tegration)	$y = 44.04 \times 10^6 x + 11.78$	$\times 10^{6}$ 0.9965	1.70 mM			
	bs-HSQC (small in	ntegration)	$y = 45.86 \times 10^6 \text{ x} - 20.802$	$\times 10^{6}$ 0.9981	1.70 mM			
	H,H COSY (2 scar	ns, 15 min)	$y = 40.58 \times 10^6 \text{ x} - 59.163$	$\times 10^{6}$ 0.9970	1.70 mM			
	H,H COSY (8 scar	ns, 57 min)	$y = 174.67 \times 10^6 x - 6.602$	$\times 10^{6}$ 0.9986	1.00 mM			
562 563 564 565 566 567 568 569 570	Table 2 Repeatability and intra-day and inter-day precision obtained from various NMR measurements. Results are given in mg/g drug material; RSD values are shown in parenthesis.							
571					_			
571	Measurement	Repeatability	Intra-day 1	ntra-day 2	Inter-day			
571	Measurement bs-HSQC _{wide}	Repeatability 15.57 (3.38%)	Intra-day 1 I 16.21 (6.35%) 16	ntra-day 2 .13 (5.45%) 16	Inter-day 5.17 (5.65%)			
571	Measurement bs-HSQC _{wide} bs-HSQC _{small}	Repeatability 15.57 (3.38%) 15.55 (3.25%)	Intra-day 1 I 16.21 (6.35%) 16 15.87 (5.73%) 16	ntra-day 2 .13 (5.45%) 16 .02 (6.00%) 15	Inter-day 5.17 (5.65%) 5.95 (5.61%)			

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578

579 Table 3

H,H COSY_{8 scans}

14.98 (1.89%)

- 580 Accuracy of various NMR measurements. Values are expressed as the percentage of recovery
- 581 (spiking experiments) and deviation from the mean value of all four experimental variants.
- 582

Measurement	High spike (120%)	Medium spike (100%)	Low spike (80%)	Deviation from the mean value
bs-HSQC _{wide}	101.2%	94.1%	89.3%	+3.0%
$bs-HSQC_{small}$	95.2%	96.2%	93.8%	+1.6%
H,H COSY _{2 scans}	95.1%	103.2%	97.6%	-1.7%
H,H COSY _{8 scans}	97.2%	98.8%	101.3%	-2.9%

15.20 (1.73%)

15.28 (1.71%)

15.24 (1.66%)

584 **References**

- 585 Akihisa, T., Hayakawa, Y., Tokuda, H., Banno, N., Shimizu, N., Suzuki, T, & Kimura, N.
- 586 (2007). Cucurbitane glycosides from the fruits of *Siraitia grosvenorii* and their inhibitory
- 587 effects on Epstein-Barr virus activation. Journal of Natural Products, 70, 783–788.
- 588 <u>https://doi.org/10.1021.np068074x</u>.
- 589 Chaturvedula, V. S. P., & Prakash, I. (2011). Cucurbitane glycosides from *Siratia grosvenorii*.
 590 *Journal of Carbohydrate Chemistry*, 30, 16–26.
 591 <u>https://doi.org/10.1080/07328303.2011.583511</u>.
- 592 Çiçek, S. S., Girreser, U., & Zidorn, C. (2018). Quantification of the total amount of black
- 593 cohosh cycloartanoids by integration of one specific ¹H NMR signal. Journal of
- 594
 Pharmaceutical
 and
 Biomedical
 Analysis,
 155,
 109–115.

 595
 https://doi.org/10.1016/j.jpba.2018.03.056.
 https://doi.org/10.1016/j.jpba.2018.03.056.
 https://doi.org/10.1016/j.jpba.2018.03.056.
 https://doi.org/10.1016/j.jpba.2018.03.056.
- 596 Çiçek, S. S., Ugolini, T., & Girreser, U. (2019). Two-dimensional qNMR of anthraquinones in
- 597 *Frangula alnus (Rhamnus frangula)* using surrogate standards and delay time adaption.
- 598 *Analytica Chimica Acta*, 1081, 131–137. <u>https://doi.org/10.1016/j.aca.2019.06.046</u>.
- 599 Çiçek, S. S. (2020). Structure-dependent activity of plant-derived sweeteners. *Molecules*, 25,
 600 1946. https://doi.org/10.3390/molecules25081946.
- Deng, F. L., Liang, X. F., Yang, L. R., Liu, Q. F., & Liu, H. Z. (2013). Analysis of mogroside
- 602 V in *Siraitia grosvenorii* with micelle-mediated cloud-point extraction. *Phytochemical* 603 *Analysis*, 24, 381–385. https://doi.org/10.1002/pca.2420.
- Fletcher, J. N., Pan, L., & Kinghorn, A.D. (2017). Medicinal chemistry of plant naturals as
 agonists/antagonists for taste receptors. In: D. Krautwurst, D. (Ed.), *Taste and Smell;*
- 606 *Topics in Medicinal Chemistry* (pp. 35–72). Heidelberg: Springer.
- 607 Girreser, U., Ugolini, T., & Çiçek, S. S. (2019). Quality control of *Aloe vera* (*Aloe barbadensis*)
- and *Aloe ferox* using band-selective quantitative heteronuclear single quantum correlation

 609
 spectroscopy
 (bs-qHSQC).
 Talanta,
 205,
 120109.

 610
 https://doi.org/10.1016/j.talanta.2019.07.004.

 </

- 611 Gong, X., Chen, N., Ren, K., Jia, J., Wei, K., Zhang, L., Lv, Y., Wang, J., & Li, M. (2019). The
- fruits of *Siraitia grosvenorii*: A review of a Chinese food-medicine. *Frontiers in Pharmacology*, 10, 1400. https://doi.org/10.3389/fphar.2019.01400.
- 614 Itkin, M., Davidovich-Rikanati, R., Cohen, S., Portnoy, V., Doron-Faigenboim, A., Oren, E.,
- 615 Freilich, S., Tzuri, G., Baranes, N., Shen, S., Petreikov, M., Sertchook, R., Ben-Dor, S.,
- 616 Gottlieb, H., Hernandez, A., Nelson, D. R., Paris, H. S., Tadmor, Y., Burgen, Y.,
- 617 Lewinsohn, E., Katzin, N., & Schaffer, A. (2016). The biosynthetic pathway of the
- 618 nonsugar, high-intensity sweetener mogroside V from *Siraitia grosvenorii*. *Proceedings of*
- 619 the National Academy of Sciences of the United States of America, 113, 7619–7628.
- 620 <u>https://doi.org/10.1073/pnas.1604828113</u>.
- Jaravine, V., Ibraghimov, I. & Orekhov, V. Y. (2006). Removal of a time barrier for highresolution multidimensional NMR spectroscopy. *Nature Methods*, 3, 605–607.
 <u>https://doi.org/10.1038/nmeth900</u>.
- Kasai, R., Nie, R.-L., Nashi, K., Ohtani, K., Zhou, J., Tao, G.-D., Tanaka, O. (1989). Sweet
 cucurbitane glycosides from fruits of *Siraitia siamensis* (chi-zi luo-han-guo), a Chinese
- 626 folk medicine. *Agricultural and Biological Chemistry*, 53, 3347–3349.
- 627 Li, C., Lin, L.-M., Sui, F., Wang, Z.-M., Huo, H.-R., Dai, L., & Jiang, T.-L. (2014). Chemistry
- and pharmacology of Siraitia grosvenorii: A review. *Chinese Journal of Natural Medicines*, 12, 89–102. https://doi.org/10.1016/S1875-5364(14)60015-7.
- 630 Li, D., Ikeda, T., Matsuoka, N., Nohara, T., Zhang, H., Sakamoto, T., & Nonaka, G. I. (2006).
- 631 Cucurbitane glycosides from unripe fruits of Lo Han Kuo (*Siraitia grosvenori*). *Chemical*
- 632 *and Pharmaceutical Bulletin*, 54, 1425–1428. <u>https://doi.org/10.1248/cpb.54.1425</u>.

- 633 Li, D., Ikeda, T., Huang, Y., Liu, D. P., Nohara, T., Sakamoto, T., & Nonaka, G. I. (2007).
- 634 Seasonal variation of mogrosides in Lo Han Kuo (*Siraitia grosvenori*) fruits. *Journal of*
- 635 *Natural Medicines*, 61, 307–312. <u>https://doi.org/10.1007/s11418-006-0130-7</u>.
- 636 Luo, Z., Shi, H., Zhang, K., Qin, X., Guo, Y., & Ma, X. (2016). Liquid chromatography with
- 637 tandem mass spectrometry method for the simultaneous determination of multiple sweet
- 638 mogrosides in the fruits of *Siraitia grosvenorii* and its marketed sweeteners. *Journal of*
- 639 *Separation Science*, 39, 4124–4135. <u>https://doi.org/10.1002/jssc.201600563</u>.
- 640 Makapugay, H. C., Nanayakkara, N. P. D., Soejarto, D. D., & Kinghorn, A. D. (1985). High-
- 641 performance liquid chromatographic analysis of the major sweet principle to Lo Han Kuo
- 642 fruits. Journal of Agricultural and Food Chemistry, 33, 348–350.
 643 https://doi.org/10.1021/jf00063a007.
- 644 Niu, B., Ke, C.-Q., Li, B.-H-, Li, Y., Yi, Y., Luo, Y., Shuai, L., Yao, S., Lin, L.-G., Li, J., &
- 645 Ye, Y. (2017). Cucurbitane glycosides from the crude extract of *Siratia grosvenorii* with
- moderate effects on PGC-1α promoter activity. *Journal of Natural Products*, 80, 1428–
 1435. <u>https://doi.org/10.1021/acs.jnatprod.6b01086</u>.
- 648 Pawar, R. S., Krynitsky, A. J., & Rader, J. I. (2016). Sweeteners from plants with emphasis
- 649 on *Stevia rebaudiana* (Bertoni) and *Siraitia grosvenorii* (Swingle). *Analytical and* 650 *Bioanalytical Chemistry*, 405, 4397–4407. https://doi.org/10.1007/s00216-012-6693-0.
- Prakash, I., & Chaturvedula, V. S. P. (2014). Additional new minor cucurbitane glycosides
 from *Siratia grosvenorii*. *Molecules*, 19, 3669–3680.
 https://doi.org/10.3390/molecules19033669.
- 654 Qing, Z.-X., Zhao, H., Tang, Q., Mo, C.-M., Huang, P., Cheng, P., Yang, P., Yang, X.-Y., Liu,
- 655 X.-B., Zheng, Y.-J., & Zeng, J.-G. (2017). Systematic identification of flavonols, flavonol
- 656 glycosides, triterpene and siraitic acid glycosides from *Siratia grosvenorii* using high-
- 657 performance liquid chromatography/quadrupole-time-of-flight mass spectrometry

- 658 combined with a screening strategy. *Journal of Pharmaceutical and Biomedical Analysis*,
- 659 138, 240–248. <u>https://doi.org/10.1016/j.jpba.2017.01.059</u>.
- 660 Schleucher, J, Schwendinger, M., Sattler, M., Schmidt, P., Schedletzky, O., Glaser, S. J.,
- 661 Sørensen, O. W. & Griesinger, C. (1994). A general enhancement scheme in heteronuclear
- 662 multidimensional NMR employing pulsed field gradients. *Journal of Biomolecular NMR*,
- 663 4, 301–306. <u>https://doi.org/10.1007/BF00175254</u>.
- 664 Shaw, A. A., Salaun, C., Dauphin, J.-F., & Ancian, B. (1996). Artifact-free PFG-enhanced 665 double-quantum-filtered COSY experiments. *Journal of Magnetic Resonance, Series A*,
- 666 120, 110–115. <u>https://doi.org/10.1006/jmra.1996.0105</u>.
- 667 Shen, Y., Lin, S., Han, C., Zhu, Z., Hou, X., Long, Z., & Xu, K. (2014). Rapid identification
- 668 and quantification of five major mogrosides in *Siraitia grosvenorii* (Luo-Han-Guo) by
- 669 high-performance liquid chromatography-tripe quadrupole linear ion trap tandem mass
- 670 spectrometry combined with microwave-assisted extraction. *Microchemical Journal*, 116,
- 671 142–150. <u>https://doi.org/10.1016/j.microc.2014.04.014</u>.
- 672 Soejarto, D. D., Addo, E. M., & Kinghorn, A. D. (2019). Highly sweet compounds of plant
- 673 origin: From ethnobotanical observations to wide utilization. *Journal of*674 *Ethnopharmacology*, 243, 112056. <u>https://doi.org/10.1016/j.jep.2019.112056</u>.
- 675 United States Pharmacopeial Convention (2018). Food Chemicals Codex (11th ed.). Rockville,
- 676 MD: The United States Pharmacopeial Convention, USP (pp. 817–818).
- 677
- 678